

Overflow- metabolism

What?

A metabolic phenomenon that is induced when the rate of glycolysis exceeds a critical value

Where?

In "glucose sensitive yeast" species: aerobic **ethanol** formation

In *E. coli*: aerobic **acetate** formation

In mammalian cell cultures: aerobic **lactate** formation

Can be avoided by process control techniques

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The Pasteur effect

"Oxygen inhibits fermentation and reduces the rate of glycolysis"

- Respiratory chain competes for ADP with PG kinase and Pyr kinase

The Crabtree effect (" **The glucose effect** ")

"High rate of glycolysis reduces the respiration and induces fermentation"

- Catabolite repression of respiratory chain enzymes

- Slow response - gene level

Overflow metabolism (" **The short-term Crabtree effect** ")

"High rate of glycolysis results in by-product formation from pyruvate

- Unknown mechanisms . Different functions in yeasts and *E. coli*

- Fast response - enzyme regulation level

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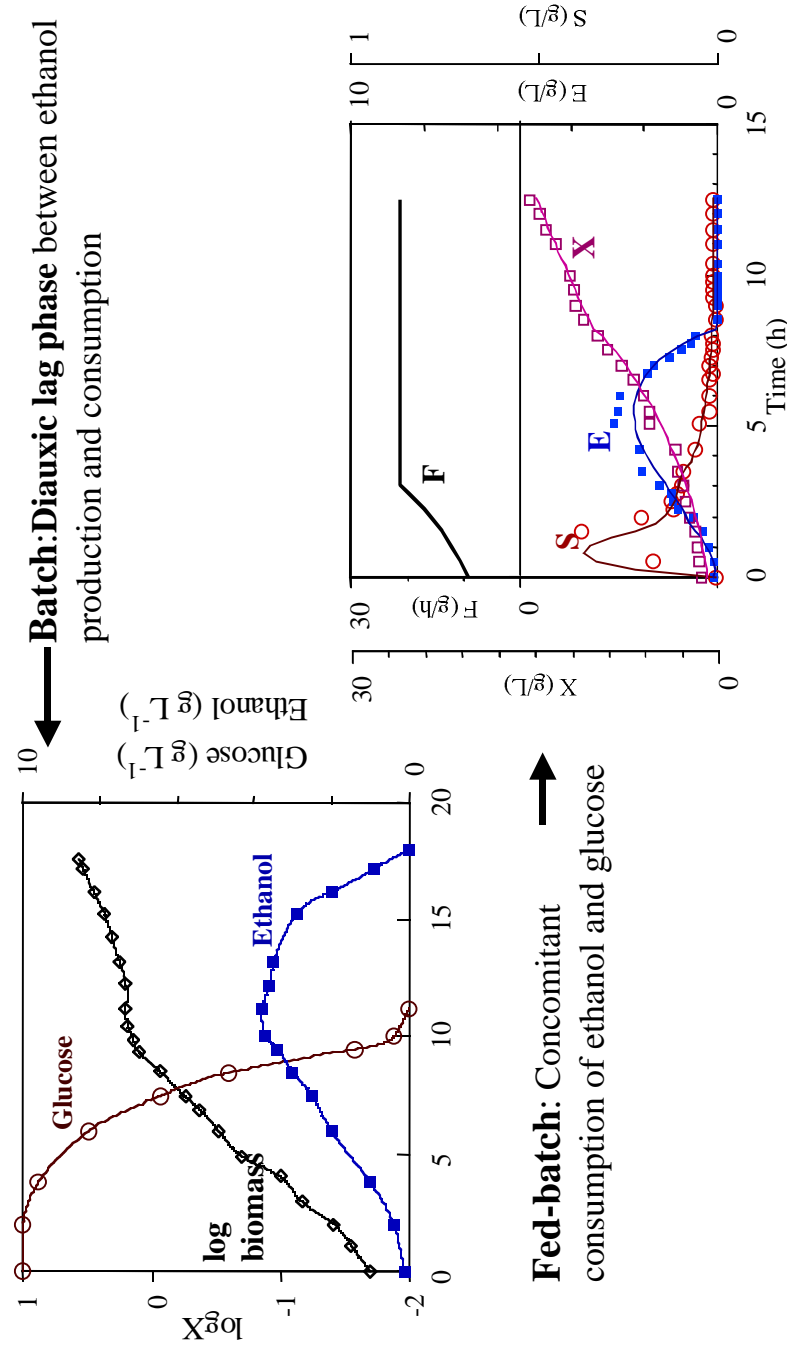
Crabtree positive yeasts:

Saccharomyces cerevisiae
S. carlbergensis
S. pasteurianus
Schizosaccharomyces pombe
Debaromyces globosus

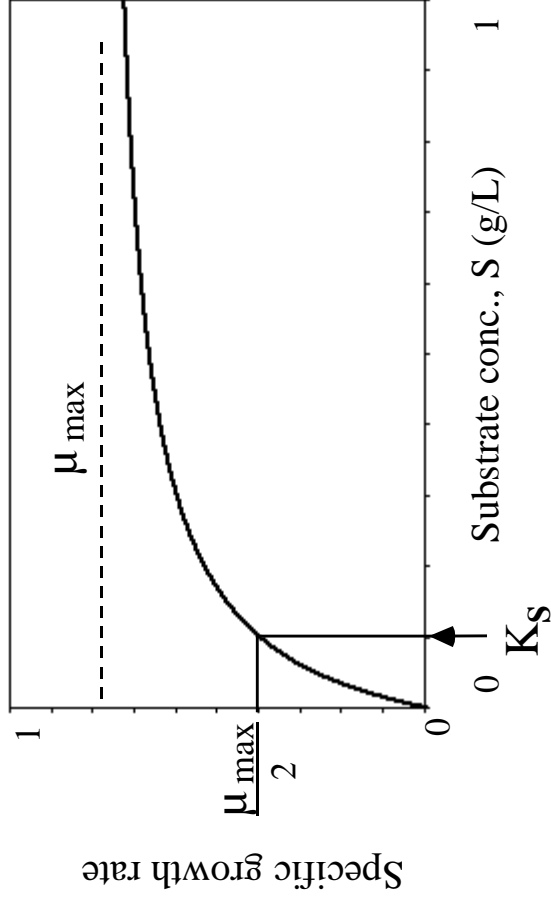
Crabtree negative yeasts:

Saccharomyces fragilis
Trichosporon fermentans
Candida utilis
C. tropicalis
Pichia fermentans

Kinetics in aerobic batch and fed-batch cultures of *S. cerevisiae*



The Monod model



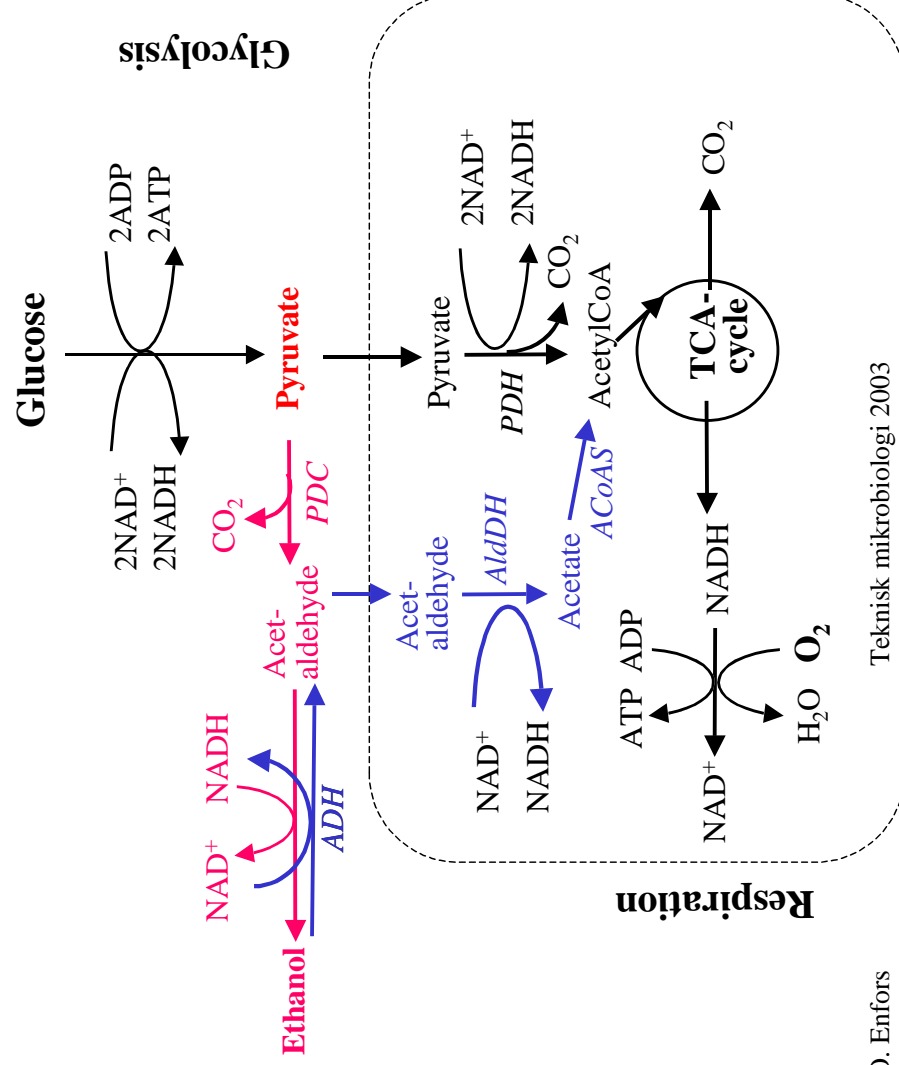
For growth:

$$\mu = \mu_{\max} \frac{S}{S + K_s}$$

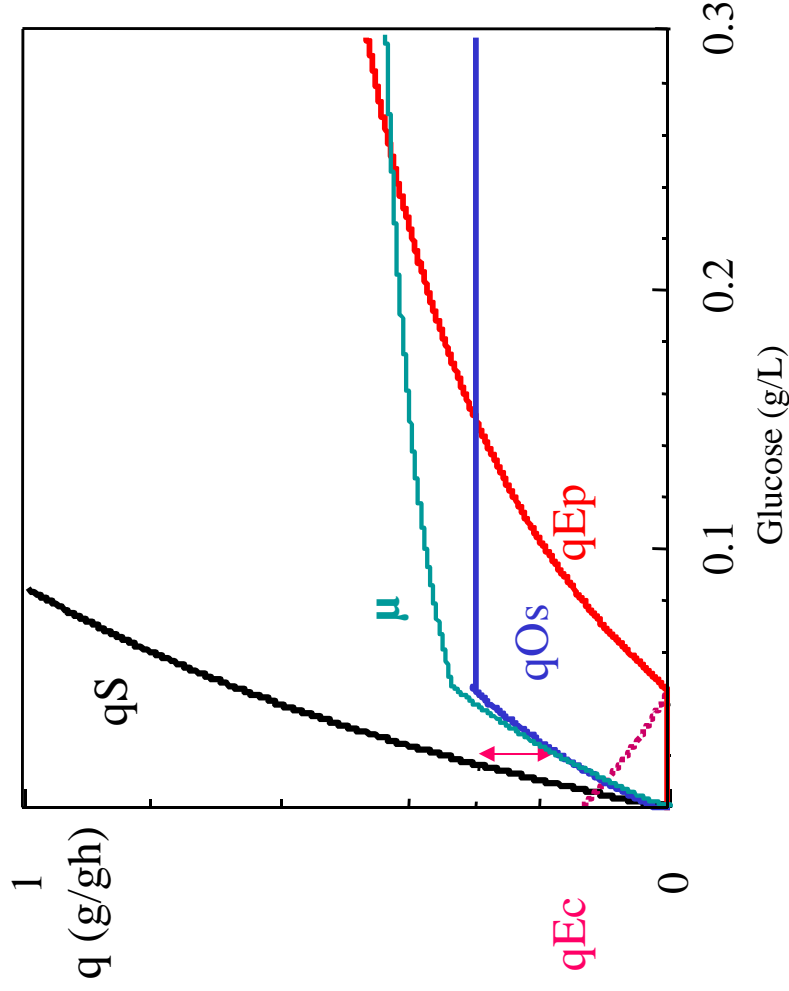
For substrate uptake:

$$q_s = q_{s\max} \frac{S}{S + K_s} = \frac{\mu}{Y_{X/S}}$$

Energy metabolism of *S. cerevisiae*



Kinetic model of yeast over-flow metabolism

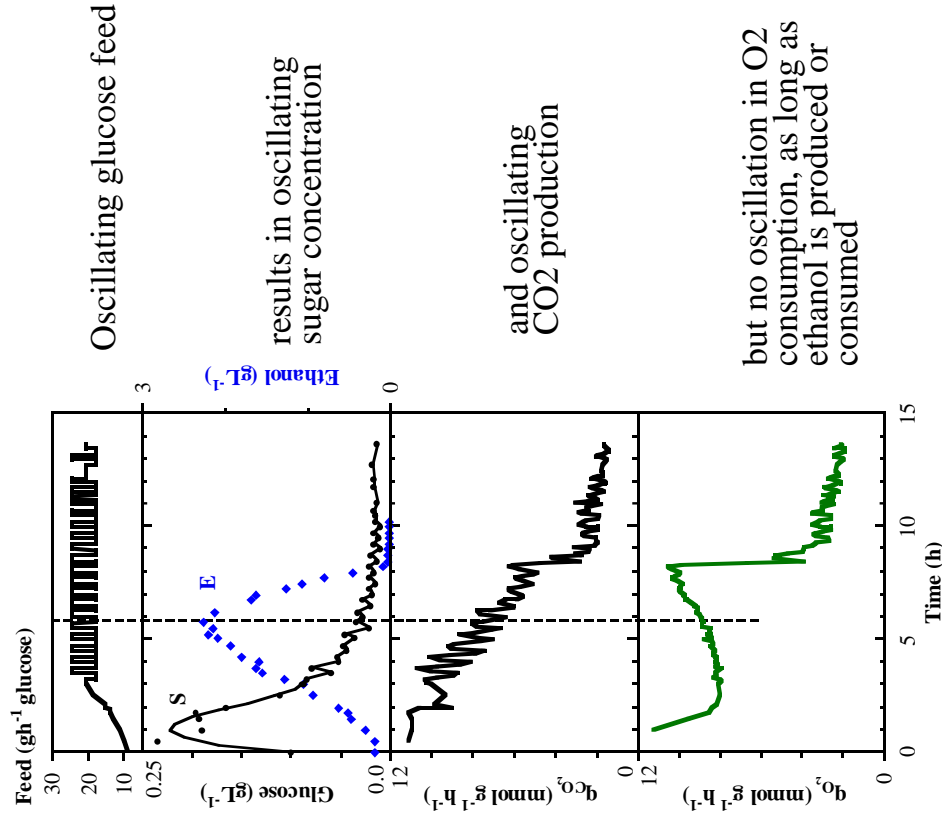


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Experimental verification of the overflow metabolism model for yeast

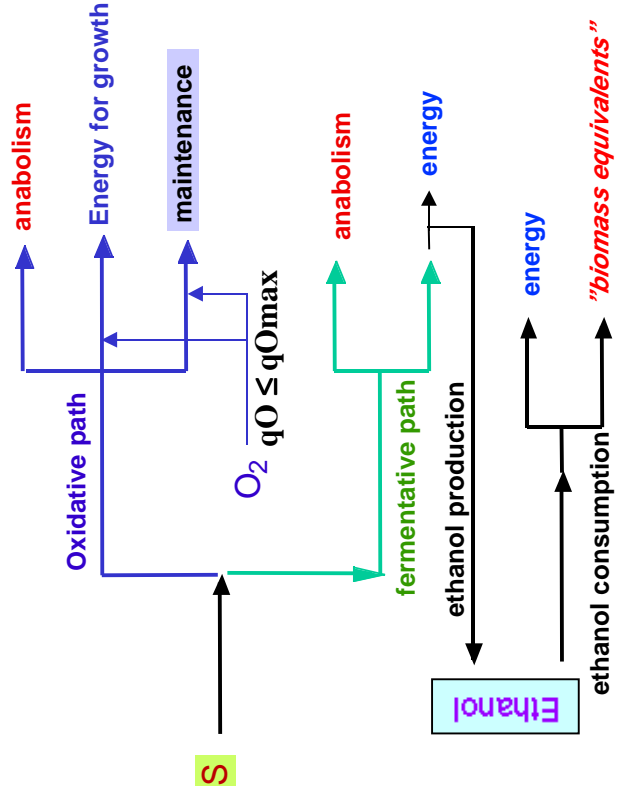


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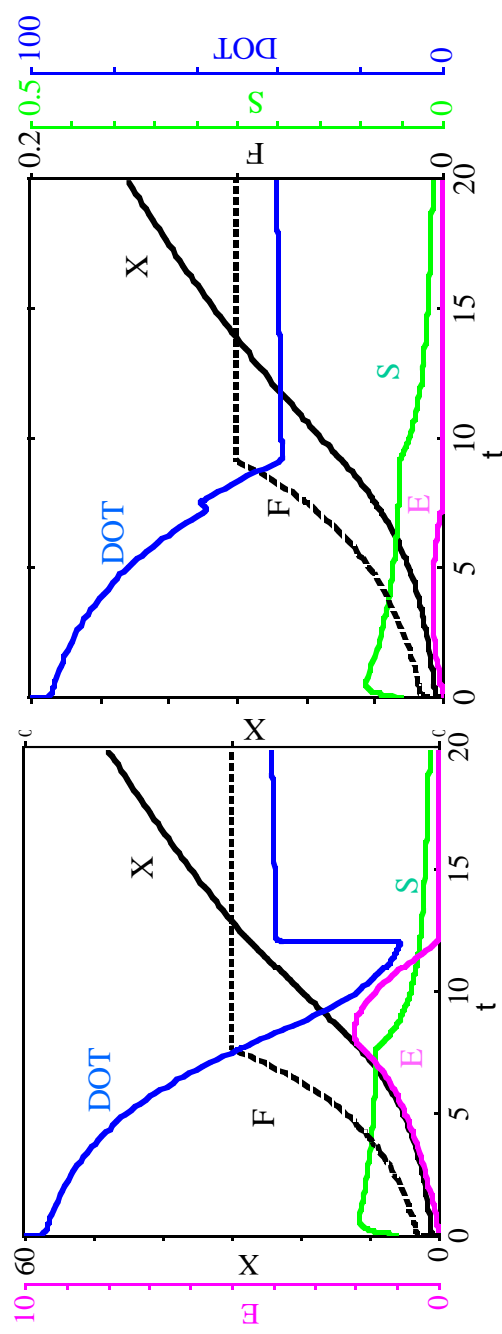
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Metabolic flux model of *S. cerevisiae*



Each arrow (flux) can be expressed as a rate and then used for process simulation

Control of over-flow metabolism by exponential feed in *S.cerevisiae* fed-batch cultures



The only difference in the two processes is that the one to the left is run with exponent 0.3 h^{-1} and the right hand process with exponent 0.25 h^{-1} which resulted in less ethanol.

